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Reveal answer. A r (relative atomic mass) of C = 12, A r of O = 16. M r (relative formula mass) of carbon dioxide = 12 + 16 + 16 = 44. number of moles = 22 ÷ 44 = 0.5 mol. Finding the mass ...

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mass of water is 18.0 g/mol and the molar mass of AlCl₃ is 133.5 g/mol. (g/mol is read grams per mole where mol is the abbreviation for mole.) Critical Thinking Questions 20. Verify using a periodic table and calculator that the molecular mass of N₂O₅ is approximately 108 amu. Show yo work below. 21. How many molecules of N₂O₅ are required to ...

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