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Relative Mass and the Mole 161 . 3. Imagine you have two baskets—one filled with quail eggs and one filled with the same number of chicken eggs. a. Which basket would be heavier? Chicken eggs. b. How many times heavier would that basket be? The basket of chicken eggs would be 16x heavier than the basket of quail eggs. c. Explain mathematically how it is possible for you to answer part b with ...

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$= 0.3333 \text{ mol} \times 6.02214 \times 10^{23}$ . Calculate the mass in grams of 2.5 moles of argon. The molar mass of argon is 39.95 g/mol.  $2.5 \text{ mol Ar} \times 39.95 \text{ g/mol Ar} = 99.875 \text{ g Ar} = 100 \text{ g Ar}$  Note 1: The units in the answer can be derived by carrying out the arithmetic operations on the units.

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this number in moles. Molar Mass Atoms and molecules are extremely small particles. Scientists have developed a method of measuring the mass of an atom relative to the mass of a chosen standard atom. This standard atom is the carbon-12 atom. The carbon -12 atom has been assigned a mass of exactly 12 amu(s). 1. What does amu stand for? 2. How is ...

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